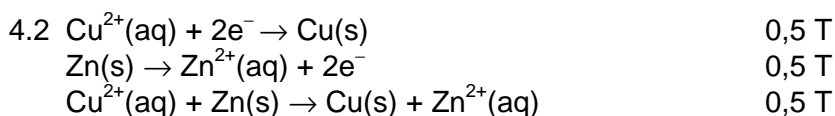
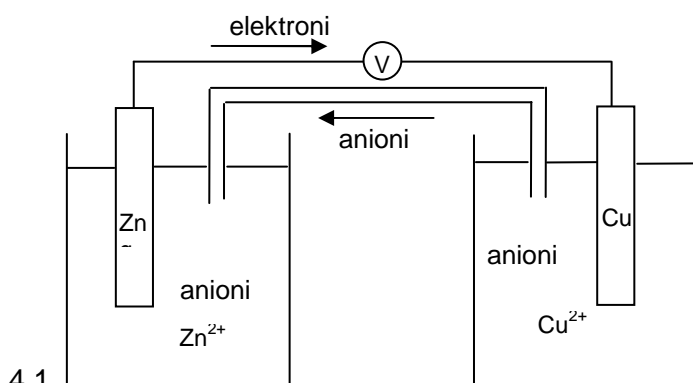
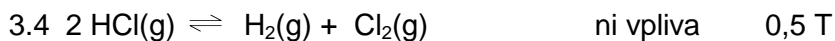
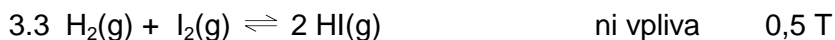
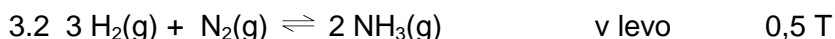
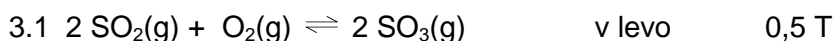
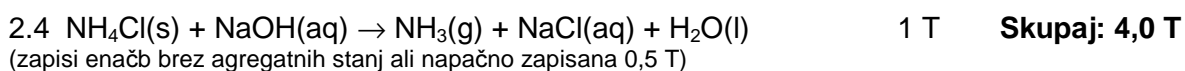
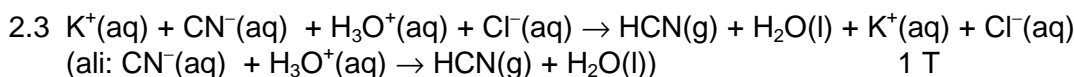
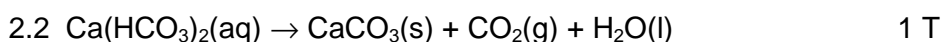
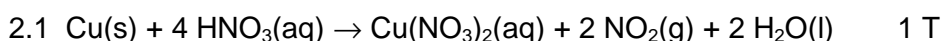
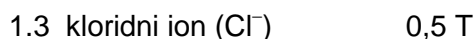
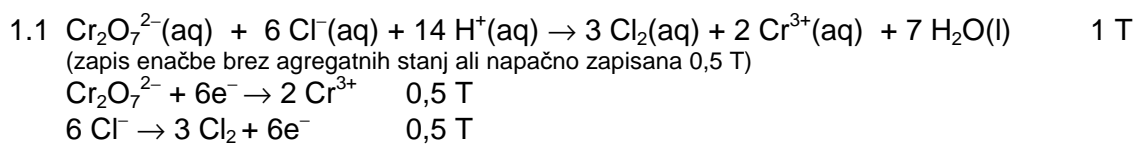


Rešitve

5.1	G ₂ Z	0,5 T	
5.2	vse	0,5 T	
5.3	A ₂ X	0,5 T	Skupaj: 1,5 T

6. A **2,0 T**

7.1	HCl(aq) + H ₂ O(l) → Cl ⁻ (aq) + H ₃ O ⁺ (aq)	1 T
	CH ₃ COOH(aq) + H ₂ O(l) ⇌ CH ₃ COO ⁻ (aq) + H ₃ O ⁺ (aq)	1 T
	(zapisa enačb brez agregatnih stanj ali napačno zapisana 0,5 T)	
7.2	V smer nastanka ionov, (ali: v smer nastanka produktov; ali v desno)	0,5 T
	V smer nastanka molekul, (ali: v smer nastanka reaktantov; ali v levo)	0,5 T
		Skupaj: 3,0 T

8.1 2 CO₂(g) ⇌ 2 CO(g) + O₂(g) 1 T

8.2 Tabela 3 T

množine/koncentracije	ogljikov dioksid	ogljikov oksid	kisik
začetna množina [mol]	2,5	0	0,2
ravnotežna množina [mol]	1,8	0,7	0,55
ravnotežna koncentracija [mol·L ⁻¹]	0,9	0,35	0,275

(za pravilno izpolnjene navpične stolpce po 1 T)

8.3 K_c = 0,04 1 T **Skupaj: 5,0 T**

9. FeSO₄ · 7 H₂O **1,5 T**

10.	A(aq) = KOH	0,5 T	
	B(g) = H ₂	0,5 T	
	C(aq) = KCl	0,5 T	
	D(l) = H ₂ O	0,5 T	
	D(g) = H ₂ O	0,5 T	
	E(s) = Cu	0,5 T	
	F(s) = CuO, Cu ₂ O	0,5 T	Skupaj: 3,5 T

Vse skupaj: 30,0 T